

**Pharmaceutical Chemistry Practice**  
**2025/2026 Academic year, 1<sup>st</sup> semester**  
**Monday 8:00-11:00**

DATE	THEME	SEMINAR, REPORT
<b>Week 1</b> 09.08.	<b>EQUIPPING, Preliminary testing</b>  Equipping, rules to be observed in the laboratory <u>Burning test</u> demonstration and discussion. <b>Practice:</b> Differentiation between inorganic and organic compounds	<u>Literature:</u> The quality control of medicinal compounds
<b>Week 2</b> 09.15.	<b>PRELIMINARY TESTING AND CLASSIFICATION OF INORGANIC AND ORGANIC COMPOUNDS</b> <b>IDENTIFICATION OF INORGANIC DRUGS I.</b>  <u>Burning and other preliminary tests</u> with model compounds. <b>Practice:</b> Preliminary tests of model compounds, their classification, identification of inorganic drugs <b>Unknown:</b> Identification of 2 inorganic drugs	Reactions of inorganic cations and anions. Preliminary tests. Classification of drugs ( <u>Literature:</u> Qualitative analytical chemistry Pharmaceutical Chemistry lecture notes Practical materials
<b>Week 3</b> 09.22.	<b>PRELIMINARY TESTING AND CLASSIFICATION OF INORGANIC AND ORGANIC COMPOUNDS, ORGANIC FUNCTIONAL GROUPS</b> <b>IDENTIFICATION OF INORGANIC DRUGS II.</b>  <u>Burning and other preliminary tests</u> with model compounds. <b>Practice:</b> Preliminary tests of model compounds, their classification, identification of inorganic drugs <b>Unknown:</b> Identification of 2 inorganic drugs, classification of 2 organic drugs	Reactions of inorganic cations and anions. Preliminary tests. <u>Literature:</u> Pharmaceutical Chemistry lecture notes Practical materials
<b>Week 4</b> 09.29.	<b>GENERAL PURITY TESTS FOR INORGANIC IONS</b>  <b>Practice:</b> Limit test reactions in Ph. Eur.: chlorides, iron, sulphates <b>Unknown:</b> <i>Natrii chloridum</i> : purity tests (appearance of solution, iron, sulphates, phosphates) <i>Borax</i> : purity tests (pH)	Theory and practice of purity tests in Ph. Eur. <u>Literature:</u> Pharmaceutical Chemistry lecture notes Practical materials

<b>Week 5</b> 10.06.	<b>DETERMINATION OF PROTONATION MACROCONSTANTS</b>  <b>pH-potentiometry:</b> - direct method: titration of ascorbic acid <b>UV - pH titration:</b> benzocaine (in small groups)  Calculation of the mole fraction of macrospecies as a function of pH (personal task)	<b>PROJECT REPORT</b> <b>(weeks: 2–4.)</b>
<b>Week 6</b> 10.13.	<b>DETERMINATION OF ORGANIC FUNCTIONAL GROUPS</b>  <b>Practice:</b> Identification of organic functional groups in model compounds <b>Unknown:</b> Classification of 2 organic molecules and determination of their functional groups	<u>Literature:</u> The quality control of medicinal compounds Pharmaceutical Chemistry lecture notes
<b>Week 7</b> 10.20.	<b>MAJOR ANALGETICS</b>  <i>Morphini hydrochloridum, Codeini hydrochloridum, Codeini phosphas, Ethylmorphini hydrochloridum, Papaverini hydrochloridum</i>  <b>Practice:</b> identification of the listed compounds  <b>Identification:</b> 1 organic compound <b>Assay:</b> 1. Codeine phosphate: acidimetry in nonaqueous medium 2. Codeine hydrochloride + papaverine hydrochloride containing powder mixture	<b>1. Midterm retake (weeks:2-4.)</b> <b>(the exact date will be discussed)</b>  Structure and chemical characteristics of the listed compounds. Nonaqueous titrations.  <u>Literature:</u> The quality control of medicinal compounds Pharmaceutical Chemistry lecture notes
<b>Week 8</b> 10.27.	<b>MINOR ANALGETICS</b>  <i>Acidum salicylicum, Natrii salicylas, Acidum acetylsalicylicum Phenazonum, Metamizolum natricum, aminofenazon Paracetamolum</i>  <b>Practice:</b> identification of the listed compounds  <b>Identification:</b> 2 organic compounds <b>Assay:</b> 1. Salicylic acid: alkalimetry 2. Phenazone: iodometry	Structure and chemical characteristics of the listed compounds.  <u>Literature:</u> The quality control of medicinal compounds Pharmaceutical Chemistry lecture notes

<b>Week 9</b> 11.03.	<b>NON-STEROIDAL ANTIINFLAMMATORY DRUGS</b> <i>Diclofenacum natricum, Ibuprofenum, Indometacinum, Phenylbutazonum, Piroxicamum</i> <b>Identification:</b> 1 organic compound <b>Assay:</b> 1. Phenylbutazone: nonaqueous titration of acids 2. Diclofenac sodium: nonaqueous titration of bases	Structure and chemical characteristics of the listed compounds. Nonaqueous titration of acids.  <u>Literature:</u> The quality control of medicinal compounds Pharmaceutical Chemistry lecture
<b>Week 10</b> 11.10.	<b>DETERMINATION of logP</b>  <b>Determination of logP:</b> - by traditional method (direct determination) - by TLC, HPLC (indirect determination)	<b>PROJECT REPORT</b> <b>(weeks: 5–9.)</b>
<b>Week 11</b> 11.17.	<b>LOCAL ANAESTHETICS</b> <i>Cocaini hydrochloridum, Benzocainum, Procaini hydrochloridum, Tetracaini hydrochloridum</i> <i>Lidocainum</i>  <b>Practice:</b> identification of the listed compounds  <b>Identification:</b> 2 organic compounds <b>Quantitative determination:</b> Nonaqueous titration of basic compounds in the presence of neutral materials: Ung. anaestheticum (lidocaine)	Structure and chemical characteristics of the listed compounds.  <u>Literature:</u> The quality control of medicinal compounds Pharmaceutical Chemistry lecture notes
<b>Week 12</b> 11.24.	<b>SEDATO-HYPNOTICS, ANXIOLITICS</b> <i>Chlorali hydras, Chlorobutanolum, Ureum, Barbitolum, Phenobarbitalum, Phenobarbitalum natricum</i> <i>Alprazolamum, Diazepamum, Medazepam, Midazolamum, Nitrazepamum</i>  <b>Practice:</b> identification of the listed compounds  <b>Identification:</b> 2 organic compounds TLC identification of benzodiazepin containing tablets <b>Assay:</b> Phenobarbital: alkalimetry, potentiometric end-point detection	<b>2. Midterm retake (weeks:5-9.)</b> <b>(the exact date will be discussed)</b>  Structure and chemical characteristics of the listed compounds.  <u>Literature:</u> The quality control of medicinal compounds Pharmaceutical Chemistry lecture notes

<b>Week 13</b> 12.01.	<p><b>DRUGS EFFECTING THE VEGETATIVE NERVOUS SYSTEM</b></p> <p><i>Pilocarpini hydrochloridum, Physostigmini salicylas, Atropini sulfas, Homatropini hydrobromidum, Homatropini methylbromidum, Hyoscini hydrobromidum, Adrenalini tartras, Ephedrini hydrochloridum, Isoprenalini hydrochloridum, Noradrenalini hydrochloridum</i></p> <p><b>Practice:</b> identification of the listed compounds</p> <p><b>Identification:</b> 1 organic compound</p> <p><b>Quantitative determination:</b></p> <ol style="list-style-type: none"> <li>1. Ephedrine hydrochloride + Codeine hydrochloride containing powder mixture</li> <li>2. Tabl. Rhinatiol Cold (ibuprofen + pseudoephedrine HCl)</li> </ol>	<p>Structure and chemical characteristics of the listed compounds.</p> <p>Nonaqueous acidimetry in multicomponent mixtures</p> <p><u>Literature:</u> The quality control of medicinal compounds Pharmaceutical Chemistry lecture notes</p>
<b>Week 14</b> 12.08.		<p><b>PROJECT REPORT</b> <b>(weeks: 10-13.)</b></p> <p><b>(The 3. midterm retake will be on the 1<sup>st</sup> week of the exam period)</b></p>

## List of inorganic compounds for identification

Acidum boricum  
Alumen  
Aluminii chloridum hexahydricum  
Aluminii sulfas  
Ammonii bromidum  
Ammonii chloridum  
Bismuthi subnitras ponderosus  
Borax  
Calcii carbonas  
Calcii chloridum hexahydricum  
Calcii hydrogenophosphas dihydricus  
Calcii sulfas  
Dinatrii phosphas dodecahydricus  
Kalii bromidum  
Kalii carbonas  
Kalii chloridum  
Kalii iodidum  
Kalii nitras  
Kalii sulfas

Magnesii chloridum hexahydricum  
Magnesii subcarbonas levis  
Magnesii sulfas heptahydricus  
Natrii bromidum  
Natrii carbonas decahydricus  
Natrii chloridum  
Natrii dihydrogenophosphas dihydricus  
Natrii hydrogenocarbonas  
Natrii iodidum  
Natrii metabisulfis  
Natrii nitris  
Natrii sulfas decahydricus  
Natrii thiosulfas  
Zinci oxidum  
Zinci chloridum  
Zinci sulfas heptahydricus