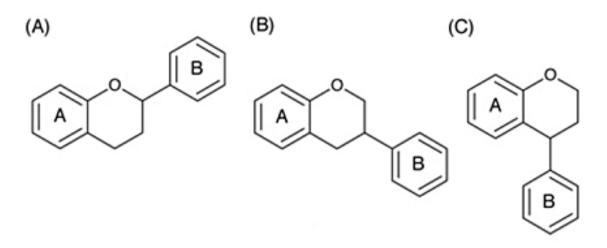
FLAVONOIDS

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Flavonoids

- Flavonoids in the broad sense of the term are virtually universal plant pigments.
- Almost always water-soluble, they are responsible for the color of flowers, fruits, and sometimes leaves,
- Examples are yellow flavonoids (chalcones, aurones, and yellow flavonols) and red, blue, or purple anthocyanins.
- When they are not directly visible, they contribute to the color by acting as **copigments**: for example, colorless flavone and flavonol copigments protect anthocyanins.
- All flavonoids (~4000) have a common biosynthetic origin, and therefore possess the same basic structural element: \rightarrow the 2- (A), or 3- (B) or 4- (C) -phenylchromane skeleton.
- Flavonoids (A), isoflavonoids (B), neoflavonoids (C).

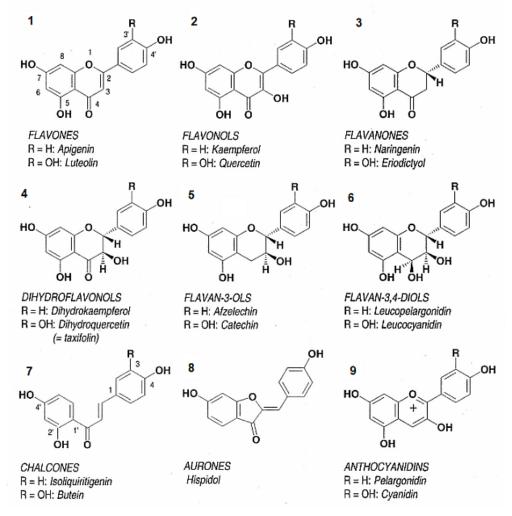


Flavonoids

• They fall into about a dozen classes depending on the degree of oxidation of the central pyran ring, which can be opened and recyclized into a furan ring (dihydrofuranone):

2-phenylbenzopyriliums (9), 2-phenylchromones (1, 2, 3, 4), 2-phenylchromanes (flavans, 5, 6); chalcones and dihydrochalcones (the pyran ring opens); 2-benzylidene coumaranones (=

aurones).



Flavones, Flavonols

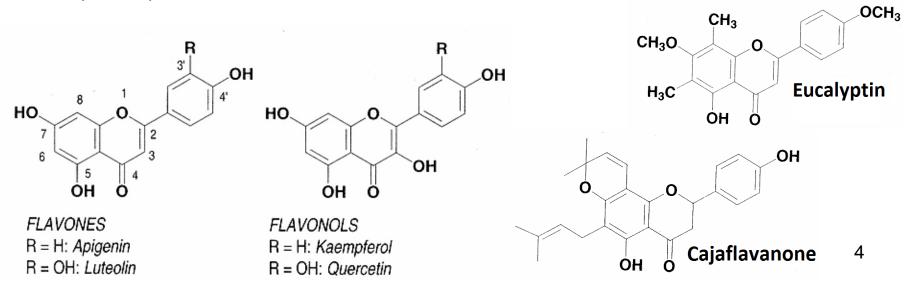
They represent the majority of known flavonoids in the strict sense.

Ring A substituents

- **Phenolic hydroxyl groups** at C-5 and C-7, (in over 90% of cases), free or etherified, or one of them engaged in a glycosidic linkage.
- Other substitutions :
 - free or etherified -OH groups at C-6 or C-8 or both,
 - isoprenylation or methylation at C-6 or C-8, or
 - involvement of C-6, C-8, or both in a carbon-carbon bond with a saccharide .

B ring

- may be substituted at C-4', or 3',4'-di-substituted, or 3',4',5'-trisubstituted; by -OH or -OCH₃
- Exceptionally, substituents at 2' and 6'can occure.



Flavanones and Dihydroflavonols

Flavanones, charcteristics:

- absence of a 2,3-double bond
- presence of at least one asymmetric center,
- C-2 is normally in the 2S configuration (natural flavanones).

Dihydroflavonols:

- Mostly 2R,3R configuration, with the phenyl and hydroxyl groups in trans (4 isomers are possible in theory).
- Structural variations are the same as those described above for flavones and flavonols.
- They are **less common** than their unsaturated homologs
- Some families tend to accumulate their C-alkylated derivatives (Asteraceae, Fabaceae).

Biflavonoids

- Flavonoids can also bond to one another, particularly through their very reactive C-6 or C-8. The
 result is a dimer known as a biflavonoid.
- Most frequently, biflavonoids are dimers of flavones and flavanones, are generally 5,7,4'trisubstituted.
- The interflavanic linkage can be
 - of the **carbon-carbon-type** (3',8", e.g., in amentoflavone; 6,8", e.g., in agathisflavone; 8,8", e.g., in cupressiflavone) or
 - of the **carbon-oxygen-carbon-type** (6-0-4", e.g., in hinokiflavone).
- The two consecutive units of the biflavonoid may or may not be of the same type (biflavone, biflavanone, flavanone, flavanone chalcone).
- The OH groups may be free or (frequently) methylated. In this group, few glycosides are known.
- Biflavonoids are **characteristic of the Gymnosperms** (see above), sporadic in the Angiosperms (e.g., Hypericum, Semecarpus, Schinus, Garcinia).

Chalcones, Aurones

Chalcones are characterized by

- the lack of a central heterocyclic nucleus, and
- a three-carbon chain with a ketone function $+ \alpha,\beta$ -unsaturation.
- Substitutions on the **A ring** are most often identical to those of other flavonoids (2',4',6'), whereas the **B ring** is fairly often unsubstituted.
- **Isoprenyl-** and **pyranochalcones** seem **rather common**, especially in the **Fabaceae**.

Aurones are characterized by a 2-benzylidene- coumaranone structure.

CHALCONES

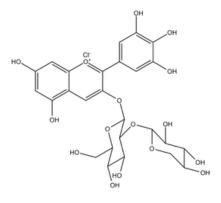
R = H: Isoliquiritigenin

R = OH: Butein

AURONES Hispidol

Glycosyflavonoids

- The **sugar moiety** may be a mono-, di-, or trisaccharide.
- Monosaccharides include
 - D-glucose, D-galactose or D-allose,
 - pentoses (D-apiose, L-arabinose, L-rhamnose, or D-xylose), or
 - D-glucuronic or D-galacturonic acid.
- **Disaccharide** and **trisaccharide moieties** may be linear or branched.



structure	usual name
O-β-D-xylosyl-(1—>2)-glucose	sambubiose
O - α -L-rhamnosyl- $(1\longrightarrow 2)$ -glucose	neohesperidose
O - α -L-rhamnosyl- $(1$ —>6)-glucose	rutinose
O-β-D-glucosyl-(1—>2)-glucose	sophorose HOW YOU
O-β-D-glucosyl-(1—>6)-glucose	gentiobiose
O-β-glucosyl-(1—>2)- O -β-glucosyl-(1—>2)-glucose	sophorotriose
O-α-rhamnosyl-(1—>2)- O -β-glucosyl-(1—>3)-glucose	2'-rhamnosyl-laminaribiose
O - α -rhamnosyl- $(1 \rightarrow 4)$ - O - $[\alpha$ -rhamnosyl- $(1 \rightarrow 6)$ -galactose]	4 ^{Gal} -rhamnosylrobinobiose

Glycosylflavonoids

(continued)

- Advances in analytical technology, especially in the field of mass spectrormetry or MS (fast atom bombardment-MS or FAB-MS, electrospray ionization MS or ESI-MS), make possible the characterization of an increasing number of acylated structures.
- Acylated structures, in which a hydroxyl group of the sugar moiety is esterified by
 - an aliphatic acid (acetic, malonic, tiglic, and others) or
 - an aromatic acid (gallic, benzoic, 4-coumaric, and other cinnamic derivatives).
- **Sulfated flavonoids** (>80 known structure).

Special Case: C-Glycosylflavonoids

- More than 300 known structure.
- The bond is established between
 - the asymmetric carbon on the sugar (often glucose, also galactose or a pentose) and
 - the C-6 or C-8 of the aglycone (often a flavone, but another type is also possible, including a flavonol or a chalcone).

Structural types:

1. mono-*C*-glycosylflavonoids (e.g., <u>scoparoside</u> in Scotch broom);

2. di-*C*-glycosyl-flavonoids (e.g., <u>isoschaftoside</u> in tea);

3. C-glycosyl-O-glycosylflavonoids (e.g., <u>saponarin</u> [= 7-O-glucosylisovitexin] in passion flower);

4. acyl- *C*-glycosyl-flavonoids (e.g., <u>4"- O-acetyl-2"-rhamosylvitexin</u> in hawthorn).

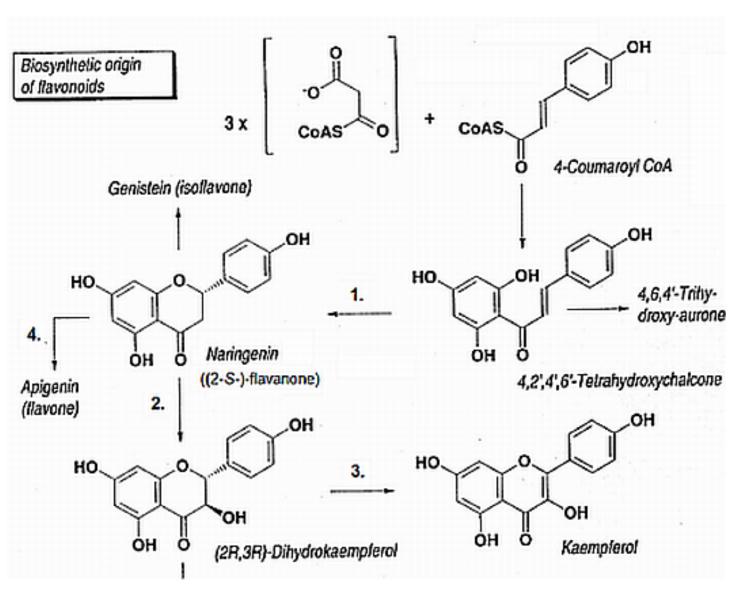
Wessely-Moser isomerization.

- In derivatives of the 5-hydroxy-C-glycosyl- flavone type, the heterocycle opens readily in acidic conditions, which explains their facile isomerization (6 <—> 8, 8 <—> 6).
- This isomerization remains of interest for the structure elucidation of these compounds.

BIOSYNTHETIC ORIGIN I.

- The key step in the formation of flavonoids is the condensation, catalyzed by chalcone synthase, of three molecules of malonyl-CoA with an ester of coenzyme A and a hydroxycinnamic acid, as a general rule 4-coumaryl-CoA.
- The reaction product is a chalcone, generally 4,2',4',6'-tetra- hydroxychalcone.

BIOSYNTHETIC ORIGIN II.



1. Chalcone isomerase, 2. Flavanone 3-hydroxylase, 3. Flavanol synthase, 4. Flavanone synthases I.& II. 12

BIOSYNTHETIC ORIGIN III.

5. Dihydroflavonol 4-reductase , 6. Leucoanthocyanidine 4-reductase

BIOLOGICAL PROPERTIES

- The main property that is recognized for flavonoids is "venoactivity", in other words their ability to decrease capillary permeability and fragility.
- In animal models, they can decrease the signs of experimental vitamin C deficiency ("vitamin P"; lack of defeciency syndrome: "vitamin P factors" → "P factors").
- Flavonoids are now referred to as "venoactive" or "vascular protective and venous tonic agents".
- The US <u>Food</u> and <u>Drug Administration</u> (= FDA) does not recognize any activity for them; their therapeutic value is generally minimized (in reference books).
- Nevertheless, flavonoids and flavonoid based preparations are—in France, Germany, Spain, Italy—widely prescribed, often recommended by pharmacists, and commonly self-prescribed to **treat** minor circulatory disorders.
- In addition, a few compounds in this class have proven efficacy, at least at high doses.
- Fields of flavonoid reserach interest:
 - interaction between flavonoids and free radicals, and its potential applications in preventive therapy,
 - activity of flavonoids on the cells and systems involved in immune responses and inflammation

Flavonoids and Free Radicals

- Biochemically, free radicals are thought to be responsible for nucleic acid alterations, mutations, initiation/promotion of carcinogenesis, and cellular damage, because of their ability to react with membrane phospholipids, among other reasons.
- Despite the lack of absolute proof, and despite the fact that their physiological role is not yet completely elucidated,
- **free radicals** are believed to be in part responsible for the genesis of atheromatous lesions, the beginning of some cancers, and neurological degeneracy.
- This has spurred research, including **epidemiologic studies** on the potential **role of antioxidants** (i.e., free radical scavengers) **flavonoids**, some lignans, other metabolites found in the daily diet, **in preventive therapy**.
- The antagonist effect towards free radical production can be studied experimentally.
 - Using experimental free radicals (hydroxyl radical, diphenylpicryihydrazyl radical),
 - radical scavenging capability can be measured in vitro (by ESR, or colorimetrically) on a lipid peroxidation model, or else
 - the activity can be evaluated *in vivo* by comparison with that of a reference antioxidant.
- Many flavonoids in the broad sense and many other phenols (especially tocopherols [= vitamin E])
 react with free radicals to prevent the degradations linked to their intense reactivity.
- It appears that the **antioxidant capability** of a flavonoid depends on its affinity for free radicals, consequently on its **structure** (in vitro, flavanols > flavonols, > flavanones, etc.).

Conclusions of two large-scale studies (from the late 1990s):

- No relationship → between incidence of cancer/mortality and flavonoid intake;
- Inverse correlation → between flavonoid intake and cardiovascular disease mortality.
 (Flavonoids protect the LDLs from oxidation, was postulated later)

Flavonoids: Enzyme Inhibitors (in vitro)

- histidine decarboxylase inhibition (by quercetin, naringenin); elastase inhibition;
- **hyaluronidase** inhibition (flavones, <u>proanthocyanidins</u>), which would maintain the intercellular ground substance in the perivascular sheath;
- non specific inhibition of catechol-O-methyltransferase, which would increase in catecholamine availability leading to a decrease in vascular fragility;
- inhibition of cAMP phosphodiesterase partly might explain → anti-platelet aggregation activity;
- inhibition of **aldose reductase** known to be involved in cataract pathogenesis— (by quercitrin, methoxyflavones, in rodents, per os);
- inhibition of **protein-kinase** (by luteolin)
- Several flavonoids—cirsiliol, hypolaetin—are potent **inhibitors of 5-lipoxygenase**, consequently are inhibitors of the production of the leukotrienes mediating inflammation and allergic reactions. Several flavonoids (luteolin, apigenin, chrysin) **inhibit cyclooxygenase** and **platelet aggregation**.

 These properties may explain anti-inflammatory and antiallergic properties
- Flavonoids: stimulants of enzymatic activity (rare):
 - **Proline hydroxylase** stimulation: would favor the formation of cross-links between collagen fibers, reinforce their strength and stability, and prevent their denaturation. This **protective activity on collagen** seems to be due mainly to **flavanol oligomers** (proanthocyanidins).
 - ! Superoxide anion radical appears to be involved in the non-enzymatic proteolysis of collagen. In vitro, anthocyanins inhibit this degradation process.

Flavonoids: Other Biological Properties

- Flavonoids are often promoted to be anti-inflammatory agents—which is compatible with what is known of their (in vitro) interactions with polymorphonuclear leucocytes, thrombocytes, or the metabolism of arachidonic acid.
- They can be antiallergic,
- hepatoprotective (e.g., isobutrin, hispidulin, flavanolignans), or
- antispasmodic (flavonoids of thyme and other Lamiaceae)
- they can decrease blood cholesterol levels, be diuretic,
- **antibacterial**, or **antiviral** in vitro (non-glycosidic 3-hydroxy-/3-methoxyflavones).
- Anticancer and tumor growth inhibitory effects in vitro:
 - they interact with the enzymes of the metabolism of xenobiotic compounds;
 - they have effects against initiation, or promotion, or both; or
 - they are cytostatic or even cytotoxic.
- Most flavonoids are **antimutagenic** *in vitro*; in contrast, some flavonols are **mutagenic** on the same models.
- The extrapolation of all of these data requires the greatest caution:
 - generally low bioavailability of flavonoids in humans,
 - frequent lack of correlation between in vitro and in vivo effects
 - the results are obtained with glycosides being probably hydrolyzed in the digestive tract
 - lack of structure-activity relationships, despite of intesive pharmacological research
 - With a few exceptions, no relevant study has shown any clinical benefit whatsoever.

CHIEF FLAVONOIDS ON THE MARKET I.

Citrus (Rutaceae)

- Trees of oriental origin, of which many species, varieties and hybrids are cultivated for their fruits and their edible endocarp.
- Widely used for their essential oils, pectins and flavonoids in the fruit pericarp.

Citroflavonoids

- mainly flavanone glycosides (hesperidin or hesperetin 7-O-rutinoside, neohesperidin, naringin, eriodyctin, eriocitrin).
- Structurally, these glycosides comprise
 - two rhamnoglycosides that differ by their linkage type—rutinose and hesperidose, 1 —>6
 and 1 —>2, respectively—and
 - 4',5,7- **trisubstituted** aglycones (naringenin, isosakuranetin) or
 - 3',4',5,7-**tetrasubstituted** aglycones (eriodictyol, hesperetin) to which
 - they are bonded through their hydroxyl group at C-7.
- The pericarps also contain **flavone glycosides** (diosmin).
- Neohesperidin and naringin are found in **bitter orange**, hesperidin (0.12-0.25 g/kg) in **sweet orange**, **grapefruit** is rich in naringin (up to 0.4 g/kg).
- Citroflavonoids are **extracted** from pericarps and pulps **with water**, and isolated using different procedures (as calcium and magnesium derivatives, by adsorption onto XAD resin).

CHIEF FLAVONOIDS ON THE MARKET II.

- Currently, the pharmaceutical industry uses:
 - a mixture of total citroflavonoids, sometimes titrated for one particular flavonoid;
 - glycosides of pure flavanones: hesperidin, naringin;
 - semisynthetic derivatives such as hesperidin methyl chalcone (the opening of the pyran heterocycle markedly increases solubility);
 - a flavone glycoside obtained by semisynthesis, diosmin.
- All of these flavonoids are used pure (diosmin, naringin) or in combinations (with one another and/or with ascorbic acid, aesculetin, ruscosides, methylaesculetin, and more).
- Accepted indications for preparations containing high doses of citroflavonoids (especially diosmin, 0.3-0.6 g/unit dose):
 - to improve the symptoms of venous and lymphatic vessel insufficiency,
 - for the adjunctive treatment of the functional signs of capillary fragility, and
 - to treat the functional symptoms of the acute attack of piles (1.2-1.8 g/day).
- The efficacy of high doses is significantly, although only slightly greater than that of a placebo, even though, in venous and lymphatic vessel insufficiency, 50% of the patients are relieved by a placebo... and by adopting elementary elements of a healthy lifestyle.
- Lower doses are proposed or used for the same indications.

Rutin: quercetin 3-O-rutinoside, Industrial sources

JAPANESE PAGODA TREE, **Sophora japonica** L., Fabaceae,

- a tall tree from central and northern China,
- cultivated in temperate climates as an ornamental species
- flower buds: 15 to 20% rutin

BUCKWHEAT, Fagopyrum esculentum Moench., F. tataricurm (L.) Gaertn., Polygonaceae

- an annual pseudocereal originated in China
- cultivated in Europe for its edible starch-containing akenes
- **leaves: 5-8% rutin**, in improved varieties

Other sources.

- the leaves of Eucalyptus macrorrhyncha F. Muell., (Myrtaceae)
- the fruits of Brazilian Caesalpiniaceae of the genus Dimorphandra.

Rutin extraction

- *S. japonica* flower buds: extraction by boiling water and crystallization upon cooling, recrystallization from ethanol.
- Fagopyrum: the presence of leaf pigments and photosensitizing substances (fagopyrins) complicate the extraction.

Rutin, Uses

Rutin, alone or in combination (with aesculin, citroflavonoids, or ascorbic acid) is promoted

- for the symptoms of venous and lymphatic vessel insufficiency,
- for the symptomatic treatment of the functional signs of capillary fragility,
- to treat the functional symptoms of the acute attack of piles, and
- for loss of visual acuity and alterations of the field of vision presumably of vascular origin.

More soluble derivatives:

- morpholinylethylrutin (ethoxazorutin, INN)
- 3',4',7 -tris-(hydroxy-ethyl)-rutin (troxerutin, INN)
- sodium rutosylpropylsulfonate
- Their uses are identical to those of rutin.

Rutin and its derivatives

They are sometimes **combined with alkaloids** (e.g., vincamine) in proprietary drugs promoted for symptomatic **treatment for senile cerebral insufficiency**.

DRUGS FOR WHICH PART OF THE ACTIVITY MAY BE DUE TO FLAVONOIDS

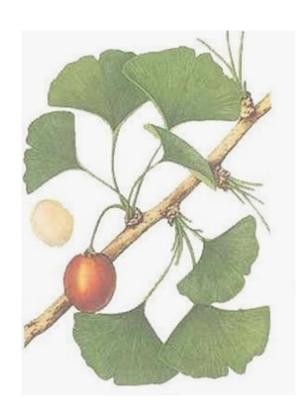
MAIDENHAIR TREE, Ginkgo biloba L., Ginkgoaceae

It is a dioecious species with deciduous leaves which originated in the Orient, the only survivor of an order widely represented until the end of the tertiary era.

- It is characterized by
- specific reproductive organs
- a "fruit" with an unpleasant odor (in reality a fertilized ovule with pulpous aril).

Cultivated

- in Korea, the southeast of France, and the USA (South Carolina)
- to supply the pharmaceutical market with leaves.
- Leaves: are commonly bilobate, can be almost entire or very divided.
- The petiole contains two bundles of conducting tissue which divide by dichotomy into the **blade**, giving it a very **characteristic striated look**.



Ginkgo folium, Chemical Composition

- sterols, aliphatic alcohols and ketones, 2-hexenal, organic acids,
- cyclitols, mono- and polysaccharides,
- flavonoids (0.5- 1%),
 - flavonol 0- glucosides (~20): quercetin and kaempferol 3-0-rhamnosides and 3-0-rutinosides,
 their 4- coumaric esters at C-6" (characteristicly with a 1"'-> 2" bond between sugars)
 - flavan-3-ols, proanthocyanidins,
 - biflavonoids: 3'—> 8" biflavones: eg, amentoflavone, bilobetol

terpene lactones

- ginkgo diterpenes: ginkgolides A, B, C, J (up to 0.5%): very specific hexacyclic structures with a spiro-[4,4]-nonanic sequence, a tert-butyl group, and three lactone rings.
- sesquiterpenes : bilobalide

$$R_1 = R_2 = H$$
: Ginkgolide A

 $R_1 = OH$, $R_2 = H$: Ginkgolide B

 $R_1 = R_2 = OH$: Ginkgolide C

 $R_1 = R_2 = OH$: Ginkgolide C

 $R_1 = R_2 = OH$: HO

 $R_1 = R_2 = OH$: Ginkgolide C

 $R_1 = H \text{ or OH}$; $R_2 = H \text{ or Glc}$:

Ginkgo biloba

Fertilized ovules

- medium-chain fatty acids (C4-C8) with a foul smell
- fleshy part: alkenyiphenols, oxidizable to quinones forming adducts with proteins and inducing cutaneous allergies.
- central almond: 4'-O-methylpyridoxine (= ginkgotoxin), may be toxic.

Ginkgo folium, Pharmacological Activity

- Ginkgolide B is an inhibitor of the platelet activating factor (= PAF), a phospholipid intercellular mediator secreted by platelets, leucocytes, macrophages, and vascular endothelial cells.
- **PAF involved processes**: platelet aggregation, thrombosis, inflammatory reaction, allergy, and bronchoconstriction (this explains the trials conducted in the late 1 990s, particularly for the treatment of asthma).
- This **anti-PAF activity** and the **activities of flavonoids**, particularly **as free radical scavengers**, may explain the numerous properties of ginkgo extract observed *in animals*.

Ginkgo extract is said to be

- a vasoregulating agent (an arterial vasodilator and a venous vasoconstrictor able to decrease capillary fragility),
- an inhibitor of cyclo-oxygenase and lipoxygenase,
- an inhibitor of platelet and erythrocyte aggregation.
- It decreases capillary hyperpermeability,
- improves tissue irrigation,
- activates cell metabolism, particularly in the cortex (by increasing glucose and oxygen uptake).
- Terpene-containing fractions: prolong the survival of hypoxic rats; they protect neurons and astrocytes from damage by transient ischemia.

Ginkgo folium, Uses

Ginkgo leaves extract: titrated to contain 24% flavonoids and 6% ginkgolides-bilobalide.

- This extract has undergone several dozen human clinical trials, especially to assess its efficacy for "cerebral insufficiency".
- "Proposed" indications (based on trials being not nanimously well received).
 - to correct the symptoms of senile cerebral insufficiency,
 - for some types of vertigo or tinnitus or both,
 - for some types of loss of hearing or loss of visual acuity (thought to be of ischemic origin)
 - for retinal insufficiency likely to be of ischemic origin.
- Recognized indication
 - the symptomatic treatment of the intermittent claudication due to chronic occlusive arterial disease of the lower limbs.
- **Side effects.** → **Oral use:** rare and minor (headaches, digestive symptoms). **Parenteral** administration: serious accidents.
- **Combinations** (e.g., with heptaminol and trihydroxy**ethylrutin** [orally], or butoforme [topically]) are indicated to treat
 - the symptoms related to venous and lymphatic vessel insufficiency,
 - the functional symptoms of the acute attack of piles.

PASSION FLOWER, Passiflora incarnata L., Passifloraceae

- The **dried aerial parts** of the passion flower are listed in the 10th edition of the French Pharmacopoeia.
- Since the constituents responsible for the sedative activity of the drug are not known with certainty, the preparations in use systematically contain the constituents as a group.
- Grows wild in the bushes of the south of the USA and Mexico.
- a creeping plant with alternate leaves with a finely serrate margin and long petioles.
- Axillary tendrils allow the plant to climb onto supports.
- The large solitary flowers (5-9 cm in diameter):
- five thick sepals, white on the underside,
- five white petals, a double crown of petaloid appendices, crimson red on the edge,
- stamens with orangy anthers,
- an unilocular ovary with three stigmas.
- The **fruit** is **ovoid** and **resembles a** small, flattened, greenish to brownish **apple** with yellow flesh.



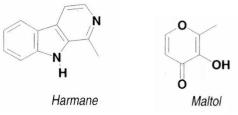


Passiflorae herba

- Passiflorae herba: fragments of lignified, hollow, grayish stem, with slender and smooth tendrils. The leaf has a long petiole and is deeply divided into three lobes with the middle one being most developed.
- The drug may be falsified with the stems and leaves of
 - P. edulis Sims. (leaf blade is dentate), and
 - P. carulea L., (pentolobate leave).

Passiflorae herba, Chemical Composition

- Phenolic acids, coumarins, phytosterols,
- I mL/kg essential oil, cyanogenic glycosides (gynocardin),
- 0.05% **malto**l (or 2-methyl-3-hydroxypyrone, maybe an artefact)
- traces of indole alkaloids: harman, harmol, harmine



30

flavonoids (up to 2.5%)

The major ones are

- flavone di-C glycosides (shaftoside and isoshaftoside = apigenin C-glucosyl-C-arabinosides = 8,6 and 6,8 isomers) and
- 2"-O-glycosides of isovitexin and iso—orientin (i.e., apigenin and luteolin C-sophorosides).

The qualitative composition can vary widely. In general, **isovitexin** and its glucosyl derivative are dominant.

R ₄		R_1	R_2	R_3	R_4
R ₂ O O O O O O O O O O O O O O O O O O O	vitexin isovitexin orientin iso-orientin saponarin schaftoside isoshaftoside vicenin-2	H glc H glc glc glc ara glc	H H H glc H H	glc H glc H ara glc glc	H OH OH H H

Passiflorae herba

Pharmacological Activity

- Tradition attributes to the passion flower **sedative**, **antispasmodic**, and "**tranquilizing**" properties partially confirmed by animal experiments (IP route).
- A body of observations supports the usefulness of "neurosedative" preparations of this drug.
- Other experiments confirm the activity of the passion flower extract on the CNS of the rat and
- point to the existence of two active compounds not yet identified.

Uses

- The drug (in infusions), its galenicals (powders, extract, tincture, nebulisate) and the phytopharmaceuticals containing it are traditionally used by the oral route
 - to treat abnormalities of the cardiac rhythm in the adult (normal heart)
 - to treat the symptoms of nervousness in adults and children, particularly minor sleeplessness
- Frequently combined with hawthorn, which has the same indication;
 with valerian and other sedative plants.
- The drug is reputed to be harmless.
- Commission E indications: mild sleeping difficulties and gastrointestinal signs of nervous origin.

THYME, Thymus vulgaris L., T. zygis L., Lamiaceae

- This Mediterranean Lamiaceae is an antibacterial and spasmolytic, but above all it is an "essential oil-containing drug".
- Spasmolytic activities are recognized for the aqueous preparations of the flower and flowering tops.
- It has been shown that
- the concentration of the volatile essential oil phenols in these preparations is insufficient to account for the spasmolytic activity,
- which is due to polymethoxyflavones and di-, tri-, and tetramethoxylated flavones, all substituted at C-6.

ROMAN CAMOMILE, Chamaemelum nobile (L.) All., Asteraceae

Roman camomile flower = the dried flower-heads of the cultivated double variety of *Chamaemelum nobile* (L.) All. [...]" (Eur. Ph., 3rd Ed.). The activity attributed to this drug is in part due to flavonoids.

- Roman camomile is a perennial plant with
- ramified stems and pinnatisect pubescent leaves of a whitish-green color.
- The capitulums of the cultivated variety have a diameter ranging from 8 to 20 mm. They practically only have ligulate flowers, white, sterile, and inserted onto a solid receptacle which bears, between the flowers, elongated and translucent paleas.
- The capitulum involucre is reduced to two to three rows of tight and imbricate bracts which are scarious on the edges.
- The ligules are lanceolate with three veins and five teeth.
- The pharmaceutical market is supplied by culture (in France and Belgium, among other countries).



ROMAN CAMOMILE, Chamaemelum nobile (L.) All., Asteraceae

Chemical Composition

- **sesquiterpene lactones** (0.6%), specifically **germacranolides**: nobilin, 3-epinobilin, and close derivatives((with a bitter taste)
- essential oil (4 to 15 mL/kg; >7 mL/kg Ph. Eur, odorous)
 - over 85% mono- and bifunctional esters of aliphatic acids and alcohols of low molecular weight (i.e., C4, C5, or C6) themselves arising from leucine, isoleucine, or valine metabolism: angelates, tiglates, methylacrylates, crotonoates, isobutanol, 3-methylbutan-1-ol, or 2-methylbutan-1-ol butyrates, and so forth.
 - monoterpenes; azulenes, only in trace amounts
- phenolic acids, coumarins,
- **flavonoids**: apigenin and luteolin glucosides.

Esters from Chamaemelum essential oil (examples)

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Pharmacological Activity

- Anti-inflammatory, antispasmodic
- may be linked to apigenin, luteolin, and their glycosides, whose activity has been established in the mouse.

ROMAN CAMOMILE, Chamaemelum nobile (L.) All., Asteraceae

Chemical Composition

- **sesquiterpene lactones** (0.6%), specifically **germacranolides**: nobilin, 3-epinobilin, and close derivatives((with a bitter taste)
- essential oil (4 to 15 mL/kg; >7 mL/kg Ph. Eur, odorous)
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 - monoterpenes; azulenes, only in trace amounts
- phenolic acids, coumarins,
- flavonoids: apigenin and luteolin glucosides.

Tests

- macro- and microscopic examination
- TLC of the flavonoid fraction of a methanol extract (spraying with aminoethanol diphenylborate).
- The proportion of capitulums of diameter smaller than 8 mm must be not more than 3%.
- Essential oil level >7 mL/kg.

Roman camomile-based medicines, Uses

Orally

- for the symptomatic treatment of gastrointestinal disorders such as epigastric bloating, impaired digestion, eructations, flatulence
- as an adjunct in the treatment of the painful component of functional digestive symptoms.

Topically

- as an emollient and itch-relieving adjunct in the treatment of skin disorders
- as a trophic protective agent for cracks, abrasions, frostbites, chaps, and insect bites;
- for eye irritation or discomfort of various etiologies (for example eye strain, seawater or swimming pool water, or smoky atmospheres);
- an antalgic in diseases of the oral cavity, oropharynx, or both (collutoria, lozenges);
- a mouthwash for oral hygiene.
- German Commission E indications: dyspepsia and inflammation of the mouth.
- Ingredient of color-lightening shampoos.

YARROW, Achillea millefolium L., Asteraceae

- Folk medicine attributes to the flowering tops of yarrow astringent, antispasmodic, and healing properties.
- The drug contains not less than 2 mL/kg essential oil and not less than 0.02% proazulenes (Eur. Ph., 3rd Ed., add. 1999).
- A.millefolium: a cosmopolitan species with sessile, deeply divided, and pubescent leaves.
- Flowers gathered in capitulums, themselves grouped in dense corymbs.
- A polymorphous species.
- Hexaploid subspecies (western Europe) :
 - ssp. *millefolium*,
 - ssp. sudetica (Opiz) Weiss
 - ssp. ceretanum Sennen (a hardy form, A. monticola Martin-Donos considered by different authors as a subspecies)



Achillea millefolium, Chemical Composition

- Sesquiterpene lactones: achillicine [= 8α-acetoxy- 10-epiartabsin], achillin, achillifolin, millefin, leukodin, dihydroparthenolide, and balchanolide
- Polyalkynes,
- Essential oil (2-10 mL/kg): azulene-containing
- **Flavonoids: luteolin**, **apigenin**, and their 7-*O*-glucosides; **flavones** and 6-methoxylated or di- and trimethylated **flavonols** (pectolinarigenin, 3-methylbetuletol, 3,6,4'-methyl- quercetagetin).
- Occurece of proazulenes yielding azulenes upon steam distillation:
 - only in tetraploid species and populations of A. millefolium in the broad sense
 - *A. collina* of central Europe is rich in proazulenes: achillicine and other derivatives of 10-epiartabsin (8α -angeloyloxy- and 8α -tigloyloxy-).
 - The essential oil of hexaploids only contains traces of azulenes.

Achillea millefolium, Pharmacological Activity and Uses

- Anti-inflammatory and antispasmodic properties are attributed, without proof, to the drug.
- These properties can be due to the flavonoids being abundant in all subspecies (in contrast to azulenes).

Yarrow is traditionally used

orally

- for the symptomatic treatment of gastrointestinal disorders: epigastric bloating, impaired digestion, eructations, flatulence,
- as an adjunct in the treatment of the painful component of functional dyspepsia.

topically,

- as an emollient and itch-relieving adjunct in the treatment of skin disorders
- as a trophic protective agent.

Commission E indications:

- internally, for gastrointestinal distress (dyspepsia, cramp-type abdominal pains, lack of appetite);
- topically (sits bath) for pelvic congestion pains in women.

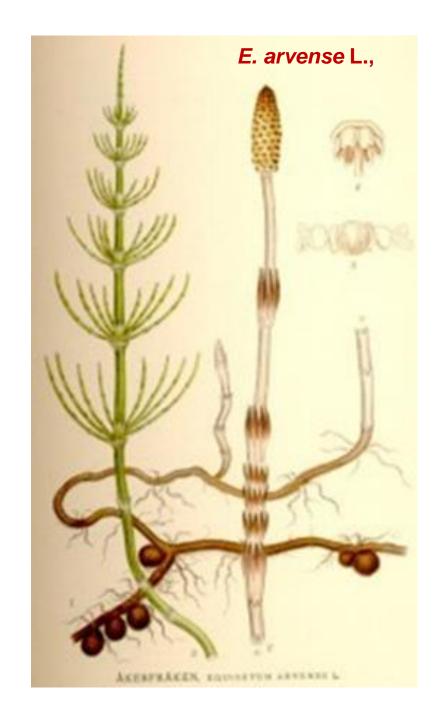
The use of yarrow is contraindicated in subjects allergic to Asteraceae because of the sesquiterpenic lactones.

FIELD HORSETAIL, Equiseti Herba, Equisetum arvense L., Equisetaceae

• Ph. Eur.: "Whole or cut, dried sterile aerial parts of *E. arvense* L."

Common in Europe, fond of moist and even swampy soils, prefers siliceous clay.

- Characterized by two types of stems: fertile stems appearing in the spring, and sterile stems following later.
- The fertile stems have no chlorophyll and support a sporangium-bearing oblong spike.
- The sterile branches (0.2-0.8 m) have hollow stems articulated at the nodes and are ribbed with 6-12 shallow grooves, and secondary branches with a quadrangular cross-section.
- At the nodes are inserted whorled leaves of very small size; shaped like black-tipped teeth and fused, they form a sheath around the stem.
- E. palustre L., a species reputed toxic (to humans?) and possible adulterant,
 - only has one type of branch with stems with six to eight deep grooves, with secondary branches with a cross-section with four or five angles, whorled in groups of eight, and hollow.
- The occurrence of alkaloids does not seem to be an absolutely distinctive characteristic.





E. palustre L.,

Equisetum stem, Equiseti herba

Chemical Composition

- **Silicon** (5-10 %): found as opaline concretions deposited onto epiderms, peripheral collenchymas, and the endoderm of stems and branches (insoluble silicon, hence the roughness of these organs).
- Sterols;
- Ascorbic acid; and phenolic acids: cinnamic acids, dicaffeoyltartaric and 5-O-caffeoylshikimic acids;
- Flavonoids (sterile stems):
 - In total, nearly 20 flavonoids in two chemotypes: <u>Asian and American</u>, and <u>European</u>.
 - Asian and American: flavone 5-O-glucosides, particularly luteolin 5-O-glucoside and its
 malonic ester at C-6" (50-60% of the total flavonoids). The other, European, is devoid of it.
 - Both chemotypes: a large amount of quercetin 3-O-malonylglucoside (ester at C-6") the major compound {30-50%} in the European chemotype), quercetin 3-O-glucoside, and other flavonol glycosides.
 - In the North Atlantic region (Scandinavia, Scotland): intermediate composition.

Equisetum stem, Equiseti herba

Tests

- TLC to show the flavonoids contained in an ethanol extract (visualization with aminoethanol diphenylborate under UV light).
- TLC detection of substitution by E. palustre:
 - carried out on the residue of acidic (H₂SO₄) extraction followed by purification by reextraction (diethyl ether, aqueous ammonia);
 - visualization with the iodoplatinate reagent must not reveal any spots.
- **HPLC analysis** allows the detection of contamination of the drug by *E. palustre*, which contains a specific flavonoid, *kaempferol 3-O-rutinosyl-7-O-glycoside*. *E. telmateia* Ehrh. is characterized by *acetylated kaempferol glycosides*.

Pharmacological Activity

- Based on tradition, and on older animal experiments, the field horsetail was considered diuretic.
- More recent experimental data only show, at best, a slight increase in water excretion.

Toxicity

Horsetails, especially E. palustre, can cause intoxications of herbivorous animals.
 Intoxication in horses is especially serious and has every aspect of an acute vitamin deficiency (loss of motor coordination).

Equisetum stem, Equiseti herba, Uses

- Horsetail-hased phytopharmaceuticals recognized indications:
 - to enhance urinary and digestive elimination functions ("traditional uses");
 - to enhance the renal excretion of water,
 - as an adjunct in weight loss programs.
- Phytotherapists frequently recommend the horsetail for bone fragility, cramps, and more.
- Contraindication: edema due to impaired cardiac or renal function
- Commission E monograph lists, among the uses for this "mild diuretic",
 - post- traumatic and static edema,
 - inflammatory or bacterial disorders of the urinary tract, and kidney stones.
 - Externally: used as an adjunct in the treatment of wounds that don't heal well.
- The horsetail and its preparations are widely used in cosmetology (prevention of wrinkles, vibex, and cellulitis).

ISOFLAVONOIDS

Isoflavonoids, I

- All molecules in this group can be related to the **skeleton of 3-phenylchromane**.
- Distribution: rather limited, isoflavonoids are almost specific to the Fabaceae.

Structural diversity:

- Structural types can be differentiated by their degree of oxidation and by the existence of added heterocycles.
- High frequency of isoprenylated derivatives, and consequently, of furan-, dihydrofuran-, and pyran-type structures.
- The most common compounds are **isoflavones**, which occur in the free state, or as glycosides (O-glycosides mainly). Related structures include isoflavanones, isoflavenes, and isoflavanes.
- **Isoflavonoids with an additional ring**: arising from the cyclization of a 2'-hydroxylated derivative: **pterocarpans** and their derivatives (pterocarpenes and 6a-hydroxypterocarpans), and also of **coumaranochromones**.

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Isoflavonoids, II

• **Isoflavonoids with a coumarinic structure** resulting from the oxidation of an isoflavene: consider the **3-arylcoumarins** (of Glycyrrhiza spp.) and the cyclization products of a 2'-hydroxy-3-arylcoumarin: **coumestans** (i.e., oxidized pterocarpans).

• **Polycyclic compounds with additional carbon atom**: e.g., **rotenoids** arising from the oxidative cyclization of a 2'-methoxyisoflavone.

- **Benzofurans**: isoflavonoids formed (maybe?) by the loss of the C-6 of coumestan.
- Isoflavonoids can form dimers and oligomers (e.g., bisisoflavanes, isoflavane-flavanone,) as well as adducts with cinnamic acids (isoflavano-lignans).

Chief types of isoflavonoids and interconversion

BIOSYNTHESIS, LIKELY ORIGIN OF ISOFLAVONES

- The mechanism proposed consists of two steps.
- The first one is the oxidation of a flavanone and its rearrangement,
- the second one is a dehydration, yielding isoflavone (e.g., (2S)- liquiritigenin —> daidzein).
- The first step is catalyzed by **isoflavone-synthase** (a cytochrome P450-dependent mono-oxydase) in the presence of NADPH and oxygen and leads to a 2-hydroxyisoflavanone.
- The mechanism is thought to involve a radical, with hydroxylation taking place at the same time as aryl migration.

ISOFLAVONOIDS, BIOLOGICAL ACTIVITY

- **Phytoalexins**: may be considered as the natural defense, mostly antifungal, products of the plants that produce them.
- The pharmacological properties of isoflavonoids are seldom known.
- **Insecticide activity** of rotenoids has found an application.
- **Estrogenic properties**: isoflavonoids cause infertility in sheep that ingest massive quantities of clover; cows seem less susceptible, perhaps because of a difference in metabolism.

Estrogenic activity of isoflavonoids

- **Isoflavones in food** → possible impact on **human health**. **In the soybean**, the concentration of **daidzein** (7,4'-dihydroxyisoflavone), **genistein** (5,7,4'- trihydroxyisoflavone) and their glycosyl derivatives (**glucosides** and **6" -O-malonylglucosides**) can reach 3 g/kg.
- These **isoflavones** and their intestinal metabolites (equol. demethylangolensin) **bind to estrogen receptors**, frequently with a **weak estrogenic activity**.
- As tyrosine-kinase inhibitors may have a role in the transformation and cell proliferation phenomena.
- Soybean-based diet: decreases mammary and prostate carcinogenesis in animal models.
- **Genistein**: an **anticarcinogen** (mammary tumors of the female rat during the neonatal period, colon microadenomas; *in vitro* result with cell culture lines).
- In humans, soybean isoflavones
 - seem to have a preventive effect on breast-, prostate-, and colorectal cancer (suggested by epidemological studies).
 - decrease the symptoms of menopause (hot flashes and others) and reduce the risk of osteoporosis.

ROTENOIDS

- These compounds, biogenetically related to isoflavonoids, have in common a **four-ring structure**: a **chromanochromanone**.
- The two oxygenated cycles are cis fused, and the biological activity is maximal for those derivatives that possess a dihydrofuran ring.
- The chief representative of the group is rotenone, the major active principle in the roots of various tropical Fabaceae of the genera Derris, Lonchocarpus, Milletia, Mundulea and Tephrosia.

Rotenone

ROTENOIDS

- *Derris elliptica* (Roxb.) Benth (Fabaceae)
 - is a vine species from Malaysia and Burma,
 - also introduced and cultivated in Africa.
 - The drug consists of the root.
 - Derris extract: enriched and titrated to contain about 30% rotenone.
 - Derris powder: with rotenoid level ranging from 3 to 10%.
- The **roots** of South American **Lonchocarpus** (**L. urucu** Killip & Smith, **L. utilis** A.C. Smith) are used in the same forms and for the same purposes.
- Rotenone has insecticidal properties;
 - active upon contact as well as by ingestion,
 - it inhibits mitochondrial respiration (NADH-dependent dehydrogenase).
- Rotenoids-containing Fabaceae (powder, extracts, rotenone) is used
 - by phytopharmacy (treatment of house plants, and, sometimes, of vegetable gardens), and
 - for the extermination of ectoparasites of domestic animals



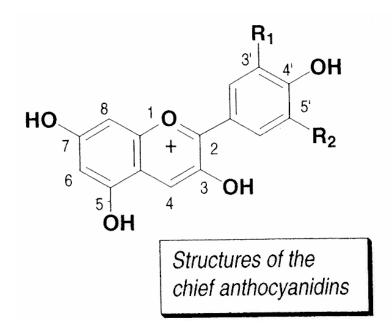
ANTHOCYANINS

Anthocyanins

Anthocyanins = antocyan glycosides, Anthocyanidins = aglycons

Aglycons

- Anthocyanidins occur in acidic medium as cations.
- They are always hydroxylated at C-3 and, most often, penta(3,5,7,3',4') or hexasubstituted (3,5,7,3',4',5') by hydroxyl groups, or methoxyl groups, or both.
- The most common aglycones (they are virtually ubiquitous) are pelargonidin (scarlet), cyanidin (crimson), and deiphinidin (purple).



$$R_1 = R_2 = H$$
: *Pelargonidin*

$$R_1 = OH$$
, $R_2 = H$: Cyanidin

$$R_1 = OCH_3$$
, $R_2 = H$: Peonidin

$$R_1 = R_2 = OH$$
: Delphinidin

$$R_1 = OCH_3$$
, $R_2 = OH$: Petunidin

$$R_1 = R_2 = CH_3$$
: Malvidin

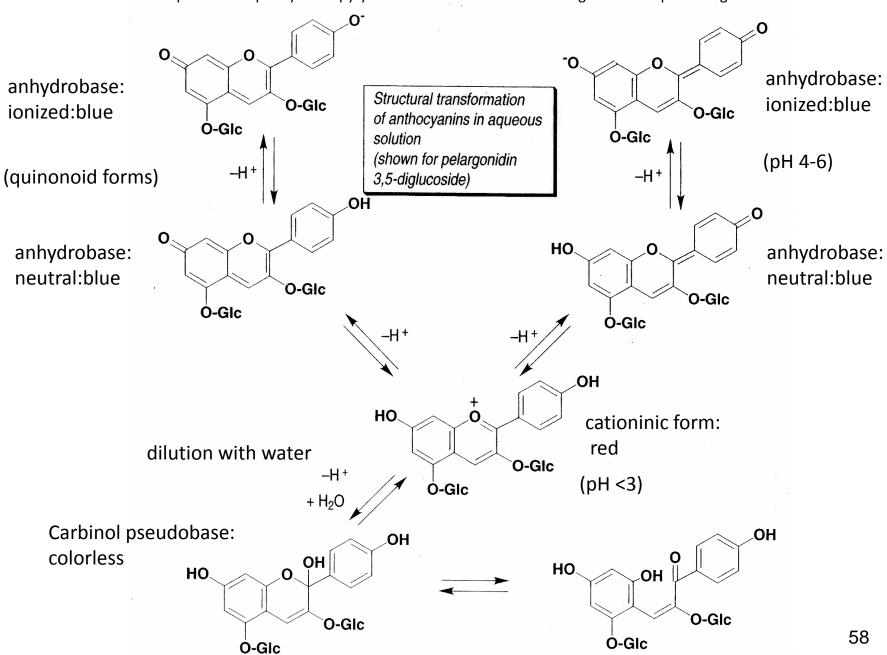
Anthocyanins

Glycosides

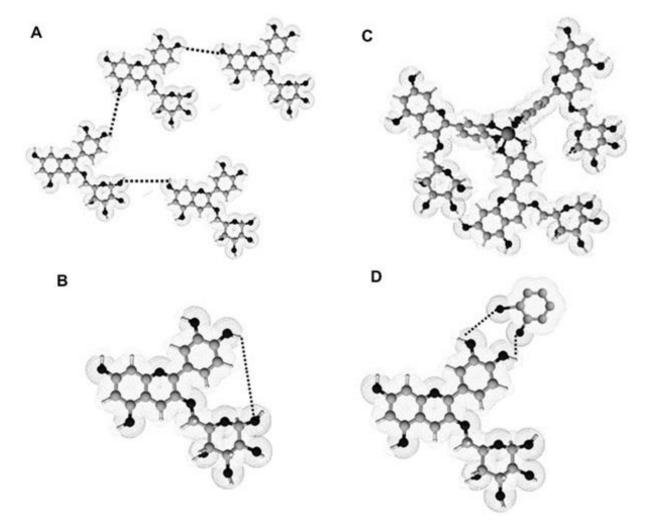
- At **least one hydroxyl group** at C-5, C-7, or C-4' **must remain free** to allow the formation of the colored quinonoid structures.
- (In contrast, anthocyanidins are unstable because their 3-hydroxylgroup makes the flavylium ion very reactive.)
- In fact, the **3-hydroxyl group is always linked to a sugar** (very often glucose) to form a stable and water soluble anthocyanin.
- 3- monosides and 3,5-diglycosides are the most common forms, 3,7-diglycosides and triglycosides (for example 3,5,3'-triglycosides) are less common.
- Sugar residue: can be a monosaccharide (e.g., glucoside, galactoside, rhamnoside), a disaccharide (e.g., rutinoside, xylosylglucoside), or, less often, a trisaccharide.
- Acylation: frequent, by phenylpropanoic acids (e.g., p-coumaric, caffeic, ferulic, sinapic acids) or benzoic acids (gallic acid), generally at C-6".; Also known by dicarboxylic aliphatic acids (e.g., malonic, malic, oxalic, succinic acids).

Biosynthetic Origin → general metabolism of flavonoids → precursors: 2,3- trans-dihydro-3,4-cis-dihydroxyflavonols → the diols undergo a hydroxylation (at C-2) and a double dehydration → glucosylation (UDP-glucose) probably occurs late

Properties of 2-phenylbenzopyrylium cation: a weak diacid and a good electrophilic reagent.



Anthocyanins interaction. (A) self-association, (B) intramolecular copigmentation, (C) metal complexation, (D) intermolecular copigmentation (Castañeda-Ovando et al.).



Co-pigmentation is a phenomenon in which the pigments and other colourless organic compounds, or metallic ions, form molecular or complex associations, generating a change or an increment in the colour intensity

Some investigations suggest that the co-pigmentation of anthocyanins with other compounds (co-pigments) is the main mechanism of stabilisation of colour in plants.

http://dx.doi.org/10.1016/j.foodchem.2008.09.001,

EXTRACTION AND CHARACTERIZATION

- Anthocyanins are soluble in water and alcohols, insoluble in apolar organic solvents, and
- They are **unstable in** neutral or alkaline medium.
- They are **generally extracted with an alcohol** (methanol, preferably ethanol if the product is intended for use in food) **in the presence of** a small amount **(0.1-1%) of hydrochloric acid**.
- To avoid esterification of the free carboxyl group of acylated anthocyanins by a diacid, and especially **to prevent** their **deacylation**, it is better to use other acids, either **weak acids** (acetic, tartaric, citric) or **volatile acids** (trifluoroacetic), or to work in a **neutral medium** (alcohol mixtures), and to work at **low temperature** (< 30 °C).
- Anthocyanin solutions are very unstable, and they can only be kept under nitrogen, at low temperature, and in the dark.
- Industrial preparation of anthocyanin extracts.
 - Classical procedure: extraction in aqueous medium containing sulfur dioxide, followed by acidification to regenerate the anthocyanins.
 - More recent procedures: ultrafiltration on cellulose membranes, chromatography on ion-exchange resins.
- **Separation of anthocyanins** is achieved by **chromatographic techniques** (column chromatography on polyamide supports, on polyvinylpyrrolidone supports, or on ion-exchange resins, preparative TLC on cellulose-coated plates, or semipreparative HPLC).

EXTRACTION AND CHARACTERIZATION

HPLC is the method of choice to analyze anthocyanin-containing drugs.

- The **separations** are most often carried out **on reverse phases with acidic water and alcohol gradients**, in which the cationic forms can be **detected at about 500-550 nm**.
- As for flavonoids, diode array detectors represent a considerable method enhancement.
- The more complex methods (**LC-MS, MS-MS**) are only available in specialized research laboratories.

As a general rule, **anthocyanin quantitation** is performed by spectrophotometry.

- At the wavelengths of maximum absorption of these compounds, interferences are exceptional:
- quantitation can be done directly on an acidic solution in alcohol (cationic form) or on an acidified juice.
- To prevent anthocyanin **self-association**, which would result in a positive deviation from the Beer-Lambert Law, **dilute solutions must be used**.

Quantitative estimates of the constituents of an anthocyanin mixture are now obtained directly by HPLC.

ANTHOCYANINS

PHARMACOLOGICAL ACTIVITY

- decrease capillary permeability and fragility (confirmed by biological tests on animals based on the diffusion of dyes)
 - participation of the collagen of the vascular wall in the control of the permeability of that wall.
 - inhibition of the proteolytic collagen degradation enzymes (elastase, collagenase); (It has been shown in vitro for black currant extracts)
- antiedema activity,
- increase in regeneration of 'visual purple' or rhodopsin (see bilberry
- act like radical scavengers in vitro (antioxidant activity).

USES

- for the symptomatic treatment of venous and lymphatic insufficiency and capillary fragility (in phlebology, proctology, or gynecology).
- in ophthalmology
 - to treat circulatory disorders of the retina or choroid,
 - to improve vision at dusk.

Proposed mechanism for the stabilisation of the cyanidine semiquinone radical (resonance) (Castañeda-Ovando et al.).

http://dx.doi.org/10.1016/j.foodchem.2008.09.001, Chemical studies of anthocyanins: A review,

ANTHOCYANINS, Other uses

The chief industrial application of anthocyanins is coloring

- they are natural pigments for which no animal toxicity has been found, be it acute or chronic.
- Sources:
 - unfermented grape juice: → liquids titrated to contain 0.5-1% anthocyanins, or
 →nebulisates titrated to contain 1-5% anthocyanins.
 - elderberry or red cabbage leaves, which are more expensive, but provide a more stable coloring agent.
- Difficulties (restricts the scope of applications):
 - instability in aqueous media (high): it results in color changes as a function of the pH, and in sensitivity to heat, light, sulfites (often used as preservatives), and metals (food cans).
 - The common occurrence of proanthocyanidins and gallotannins in the extracts can also be a problem (for example it makes gelatin precipitate in jams).
 - insolubility of anthocyanins in lipids.
- Anthocyanins are extracted from edible fruits and vegetables and may be used as food additives
 (Eur. id. code E163), for example in beverages (30 mg/L), jams, and confectionery products, to
 name only a few.

CHIEF ANTHOCYANIN-CONTAINING DRUGS

BILBERRY, Vaccinium myrtillus L., Ericaceae

- The bilberry is a subshrub with coriaceous leaves.
- The **bell-shaped flowers** grow solitary or in pairs at the base of the leaves.
- The fruit is a multiseeded tetra- or pentalocular globose berry with a fleshy mesocarp;
- on the flattened top, the remains of the style and the calyx form a **small disc with a dull edge**.
- Blueberries are particularly abundant in the woods that grow on siliceous soils in the mountains of the northern hemisphere.
- The French market is largely dominated by imports (from Poland).
- Other species (e.g., V. corymbosum, cultivated in Germany) are also used in the food industry.





Vaccinium myrtillus, Chemical Composition

Blueberries

- Water (up to 90%), sugars (3 to 7%), and organic acids.
- Phenolic acids, **flavonoids** (hyperin = hyperoside], quercitrin), **proanthocyanidins** (procyanidins B-1 and B-4), and monomeric flavan-3-ols (catechin and epicatechin).
- **Anthocyanins** (in the **fresh fruits**: about 0.5%): C-3 *O*-glucosides, *O*-galactosides, and *O*-arabinosides of cyanidin, peonidin, delphinidin, malvidin, and petunidin.

Bilberry leaf

- Phenolic acids, flavonoids (rhamnoglucosyl-, arabinosyl-, and glucuronylquercetin).
- Traces of quinolizidine alkaloids (myrtine, epimyrtine).
- <u>Proanthocyanidins</u> and catechin (<u>up to 10 %)</u>.

Uses

- Water –soluble bilberry powder titrated to contain 70 % anthocyanins. It is an ingredient of drugs used to treat
 - the functional symptomes of venous and limphatyc vessel insufficiency ,
 - cutaneous capillary fragility, and
 - mesopic and scotopic vision (nyctalopia, myopia).

Vaccinium myrtillus, Uses, Indications

In France

- Bilberry fruit, fresh or dried, and bilberry leaf: traditionally used
 - to treat the subjective symptoms of venous insufficiency, such as fullness in the legs, and
 - to relieve the symptoms of piles.
- **The fruit, fresh or dried**: is traditionally used for the adjunctive therapy of the painful component of **functional dyspepsia**.
- Only the fresh fruit: for the symptomatic treatment of the functional symptoms of capillary fragility,
- Bilberry leaf or dried fruit: for the symptomatic treatment of mild diarrhea.

In Germany, Commission E:

Bilberry fruit:

as an **adstringent**, in case of **diarrhae**;

as a topical **anti-inflammatory** in case of **irritation** of the **mucous membranes** of the mouth and throat.

CRANBERRY, Vaccinium macrocarpon Aiton, Ericaceae

- Cranberry grows wild in eastern North America, from the Carolinas to Canada.
- Cultivated in the United States since the beginning of the nineteenth century,
- It produces small dark red fruits
- widely consumed as such (**fresh or frozen**) and as cranberry juice (pure or as a cocktail sweetened with corn syrup), cranberry **sauce**, and so on
- The fresh fruit is very rich in acids (citric. quinic. benzoic);

• it also contains **anthocyanins** (3-O-galactosides and 3-O-arabinosides of cyanidin and peonidin),

• catechin, and flavonoids.





Vaccinium macrocarpon

- The beneficial—bacteriostatic-—-effect of cranberry juice in the treatment of urinary infections is confirmed by secular use.
- It is now postulated that the activity is due to the **inhibition of bacterial adhesion onto** mucous membranes.
- This has been demonstrated in the case of E. coli adhesion onto urinary tract epithelial cells, using cranberry juice as well as the urine of mice or humans collected after administration of cranberries.
- The active constituent (possibly a procyanidine) inhibiting the adhesins specific to the pathogenic strains of E. coli has since been isolated from cranberry juice—and also from bilberry (blueberry) juice.
- More recently, a placebo-controlled, double-blind clinical trial showed that
- the daily consumption of 300 mL of a commercially- available cranberry juice induced, in elderly women (average age 78.5 years), a very significant decrease in the frequency of urinary bacterial contamination, after 4-8 weeks of treatment, a delay which may correspond to an initial action on the intestinal bacterial flora.
- In the United States, the recommended use of cranberries is as a dilute juice; dried juice capsules are also available.

BLACK CURRANT, Ribes nigrum L., Grossulariaceae

- This bushy shrub is cultivated for its edible fruits (in the Burgundy region of France and in central Europe).
- The black currant leaf as well as the fruit are used in pharmacy (Fr. Ph., 10th Ed.).
- Tri- to pentalobate leaves, the underside of which are pale, pubescent, and scattered with yellow secretory glands.
- Flowers: reddish, grouped in dangling racemes, a pubescent calyx which is longer than the corolla.
- Fruit: a fragrant black berry on top of which the remains of the calyx can be seen.



Constituents

- Fruit: sugars (10-15%); organic acids; flavonol glycosides; anthocyanins: cyanidin and deiphinidin glycosides.
- **Leaves:** a small amount of essential oil; flavonoids: hyperin, astragalin, rhamnoglucosides and glucoxylosides of quercetin and kaempferol; dimeric and trimeric prodelphinidins.

BLACK CURRANT,

Uses

- Fruit
 - to prepare extracts enriched in anthocyanins
 - With therapeutic indications identical to those of the bilberry
- Leaves (in phytopharmaceuticals, traditionally)
 - to facilitate urinary and digestive elimination functions,
 - to enhance the renal excretion of water, and
 - as an adjunct in weight loss programs.
 - orally and topically, for the symptomatic treatment of minor painful symptoms of the joints.

Contemporary phytotherapy prescribes the *preparations based on the* **buds** in the same fashion.

- These are rich in diterpenoid acids (hardwickiic acid), and are
- prized for their **essential oil**, which is used in food technology.
- **Essential oil composition** varies with cultivars, but the **chief constituents** are almost always **hydrocarbons** (A3- carene, sabinene, phellandrenes, and limonene).

VINE, Vitis vinifera L. (tinctoria varieties), Vitaceae

• The term "vine" designates cultivars with black grapes, red pulp, and leaves that turn red in the fall, partially or completely.

- The dried vine leaf was the subject of a monograph in the 10th edition of the French Pharmacopoeia.
- The pharmaceutical industry also uses grape seeds.



Constituents

- Anthocyanins (up to 0.3%): 3-O-glycosides of cyanidin and peonidin;
- monocaffeoyltartaric acid, phenylpropanoic acids,
- flavonol glucosides,
- hydrolyzable tannins (esters of glucose and of gallic and dehydrohexahydroxydiphenic acids),
- proanthocyanidins.

VINE

Uses

Vine leaf-based phytopharmaceuticals are traditionally used (orally and topically) to treat

- the functional symptoms of capillary fragility such as ecchymosis and petechiae,
- the subjective symptoms of venous insufficiency such as fullness in the legs, and
- the symptoms of hemorrhoids.
- Topically, they are traditionally used for eye irritation or discomfort of various etiologies (e.g., eye strain, seawater or swimming pool water, or smoky atmospheres).

EUROPEAN ELDER, Sambucus nigra L., Caprifoliaceae

- The flower—it is the subject of a monograph in the European Pharmacopoeia.
- The fruit, a source of extracts used as food coloring.
- The European elder is a shrub widespread in western Europe.
- Its bark has small cracks and
- its leaves are imparipinnate.
- Large (20 cm) inflorescences of strong-smelling flowers,
- black berries with their purplish-red juice and three seeds.
- The flower is fairly easy to identify.
- However, to verify the absence, in the drug, of flowers of dwarf elder (S. ebulus L)—with red instead of yellow anthers—the French Pharmacopoeia requires a TLC analysis of the flavonoid content of a methanol extract.





EUROPEAN ELDER

Flowers

Constituents

- Flavonoids (>0.8%, Ph. Eur.): rutin, isoquercitrin,
- Derivatives of caffeic acid, free and esterified.
- Triterpenes
- Essential oil
 - smells like muscat grapes, has a pasty consistency,
 - contains fatty acids, 3,7-dimethyl-1,3,7-octatrien-3-ol, linalol, cis-hexenol, and rose oxides, among others.

Uses

Elder flower

- In France it is traditionally used
 - to enhance the urinary and digestive elimination functions,
 - as an adjunct in weight loss programs, and
 - to enhance the renal elimination of water
- In Germany, Comission E indications
 - As a sudorific, causing an increase in bronchial secretion
 - Colds and coughs

EUROPEAN ELDER

Fruits

Constituents

- Cyanidin glycosides: 3-O-glucosyl-, 3-O-sambubiosyl-, 3,5-diglucosyl-, and 3-sambubiosyl-5-glucosyl-cyanidins.
- Flavonoids, acids (citric, malic), saccharides, 0.1 mL/kg essential oil.
- The seed contain cyanogenic glycosides .
- The ripe fruit,
 - edible fresh or as a jam ,
 - is the source of an extract used as food coloring (e.g., to color cherry or pomegranate syrup)
- Traditionally used in France for medicinal purposes, for the same indications as the leaf.

Phenoloids in Zingiberaceae family (diarylheptanoids and arylalkanones)

TURMERIC, Curcuma domestica Val. = C. longa L., Zingiberaceae

- Perennial by a rhizome,
- turmeric has large sheathing leaves with an elliptic blade and pinnate veins.
- The flowers are yellow, gathered into a spike with bracts, and
- have an irregular corolla with a developed posterior petal,
- an androecium reduced to one fertile stamen and staminodes forming a petaloid label, and
- a gynoecium with three carpels.



- Several cultivars grown in India, Sri Lanka, Indonesia, China, and Jamaica. For the most part (80%) the world production comes from India.
- Commercial turmeric commonly consists of the ovate primary rhizomes ("bulb" or "round" turmeric), the cylindrical secondary rhizomes ("fingers"), or a mixture of both.
- Fingers: gray and grooved surface and a diameter of about 1 cm.
- Break with a clean fracture, reddish-yellow inside; odor aromatic, taste warm, somewhat bitter.

TURMERIC

Chemical Composition

- Starch (45-55%); arabinogalactans (ukonans)
- **Essential oil** (2.5 to 6%) rich in monocyclic sesquiterpenes :
 - hydrocarbons: zingiberene, and 6-curcumene; their oxygenated derivatives:
 turmerone, S-(+)-ar-turmerone, curlone, α- and y-atlantone;
- **Curcuminoids** (can reach 8 %):
 - The coloring principles in the drug, structurally related to a diarylheptane,
 - curcumin (50 to 60%), desmethoxycurcumin, bisdesmethoxycurcumin, dihydrocurcumin.

TURMERIC, Pharmacological Properties

Curcumin

- Its **anti-inflammatory activity** has been demonstrated in animal experiments and promoted by observations reported in India in man.
- Is apparently devoid of side effects.
- The mechanism of action remains poorly-understood:
 - inhibition of the increase in activity of lysosomal enzymes,
 - effect on the synthesis of prostaglandins, or
 - interference with the response of granulocytes to stimuli linked to the inflammatory phenomenon.

The drug has a definite action on the

- Hepatic parenchyma: the hydroalcoholic extract prevents the cytotoxic effects of carbon tetrachloride in vivo in the mouse and in vitro in cultured rat hepatocytes.
- Stomach: the ethanolic extract (0.5 g/kg in the rat) is active against ulcers and protects cells.

TURMERIC

Uses

- **Food coloring:** turmeric cultivars with the highest curcumin content (e.g., Allepey, > 6.5%).
- Curcumin (>90%):
 - It is a nontoxic authorized color (Eur. id. code E100).
 - It is heat resistant and scarcely sensitive to changes in pH.
 - It is used as the rhizome powder, or the oleoresin, or extracts and curcumin solutions of variable concentration, sometimes adsorbed onto hydrocolloids.
- **Spice:** Madras (3.5% curcumin) and other cultivars. **Turmeric** is, alongside coriander and other spices, **one of the main ingredients** of **curry powders** (these may also contain chili, ginger, clove, fenugreek).
- Oleoresin is also used in food technology.
- In phytopharmaceuticals: traditionally used
 - as a choleretic and cholagogue,
 - for functional dyspepsia attributed to a hepatic origin,
 - as an appetite stimulant.
 - Biliary tract obstruction is a contraindication (Commission E).

TEMU LAWAQ, *Curcuma xanthorrhiza* Roxb., Zingiberaceae

- Temu lawag is botanically very close to turmeric, and is a cultivated **Indonesian species.**
- The rhizome is cut after being harvested, so the drug appears as thin round slices.

Constituents

- Starch (30-40%),
- **Essential oil** (up to 12%): rich in **sesquiterpenes**: **zingiberene**, **ar-curcumene**, (R)(-)-xanthorrhizol, turmerones, bisacurones, bisacurol, and bisacurol.
- Curcuminoids (1-2%): curcumin, its monodemethoxylated derivative, and its di-, hexa-, and octahydrogenated derivatives. Monophenolic and non-phenolic analogs have been isolated from rhizomes collected in Thailand.

Uses

- A traditional folk remedy in southeast Asia: it is used as cholagogue and choleretic.
- In Germany, Commission E :
 - its use acceptable for gastrointestinal symptoms,
 - biliary tract obstruction is a contraindication,
 - prolonged use can cause gastric irritation.

GINGER, Zingiber officinalis Roscoe, Zingiberaceae

This **spice**, is used in the oriental traditional medicines, especially for **functional dyspepsia**.

- Originally from India, ginger is cultivated in India, China, and all of southeast Asia (Indonesia, Philippines), and in the tropical regions of Africa (Nigeria).
- Large herbaceous perennial plant,
- lanceolate leaves,
- thick inflorescence with overlapping lateral bracts,
- pale green flowers with purple label.



- The rhizome is ramified within one plane.
- Appearance, depending on the mode of preparation: gray with a wrinkled surface (coated or unscraped), white with a smooth surface (uncoated or scraped), or prepared (preserved).
- Fibrous and granular fracture; odor aromatic, taste warm and pungent.



GINGER

Chemical Composition

- Starch (60%), proteins, fats (10%), from 10 to 25 mL/kg essential oil, and a resin.
- Essential oil (composition highly depends on geographical origin):
 - **Sesquiterpene hydrocarbons** (30-70%): (—)-**zingiberene**, (+)-**ar-curcumene**, (—)-β-sesquiphellandrene, E,E-β-farnesene, β-bisabolene. Monoterpene aldehydes (citrals) and alcohols.
- **Gingerols =** 1 -(3' -methoxy-4' -hydroxyphenyl)-5-hydroxyalkan-3-ones:
 - [3-6]-, [8]-, [10]-, and [12]-gingerols bearing a side chain with 7-10, 12, 14, or 16 carbon atoms, respectively;
 - alongside the corresponding ketones, and dehydration products (shogaols).
- Labdane-type diterpenes, galonolactone and its dialdehyde derivative.

GINGER, Pharmacological Properties

Used since remote times in India and China.

• Animal experiments :

- oleoresin is a cholesterol lowering agent (in rodents),
- [6]-gingerol is a cholagogue (in the rat by the intraperitoneal route),
- [8]-gingerol is a hepatoprotective agent (prevents the toxic effects of carbon tetrachloride in rat hepatocytes).
- the acetone extract and zingiberene have an antiulcer effect in the rat
- The drug has an anti-inflammatory activity (possibly acting on prostaglandin and leukotriene production).
- Human studies (on antiemetic properties)
 - Most trials reveal an activity superior to that of a placebo for motion sickness, postoperative nausea, or morning sickness (at the usual dose of 1 g per day).
 - These trial results are divergent; (among the reasons: ginger products used were not standardized).
- Antiemetic action: may be the consequence of direct effects on the gastrointestinal tract: in the mouse, the stimulation of gastrointestinal motility by the acetone extract (75 mg/kg), by [61-shogaol (2.5 mg/kg), or by gingerols is comparable to that of metoclopramide (10 mg/kg).
- Other authors, however, noted the lack of effect of ginger powder on the rate of gastric emptying in healthy humans.
- The drug is not toxic and has no side effects.

GINGER

Uses

- Used (especially for functional dispepsia) for over 25 centuries in the formulation of countless traditional Oriental remedies (China, Japan).
- In France, in phytomedicines: traditionally used for motion sickness.
- In Germany: the rhizome powder is used for gastrointestinal distress and to prevent motion sickness (2 g/day).
- Commission E:
 - Ginger is a spasmolytic in animals
 - in humans, it has antiemetic, positive inotropic, and stimulant effects (intestinal peristalsis, salivary and gastric secretions).
 - Ginger must not be used to prevent morning sickness in pregnant women.

KAVA, Piper methysticum Forst. f., Piperaceae

• *P. methysticum,* a pepper tree which grows in the islands of western Polynesia (Papua New Guinea, Tonga, Samoa, Fiji, Vanuatu) and as far as Tahiti.

- A perennial dioecious subshrub with
- cordate leaves.
- Decaploid and sterile.
- Multiplies by vegetative propagation.



- The term **kava** designates **a beverage** prepared by soaking in water the rhizome or root fragments, after grinding them with a pestle or chewing them.
- It has been consumed for centuries according to a ceremonial described in 1875 by Captain Cook.
- This ritual beverage induces a sensation of well-being.
- It continues to play an important role in the culture of that part of the globe.

KAVA

Constituents

- Mono- or di-**unsaturated α-pyrones**, substituted by a styryl or phenethyl group, itself substituted (methoxyl, methylenedioxyl) or not.
 - They include yangonin, (+)-methysticin, (+)-dihydromethysticin, (+)-kawain, (+)-dihydrokawain, demethoxyyangonin, and minor products (e.g., dehydrokawain, 7,8-dihydroyangonin, 10- and 11-methoxyyangonins).
- **Resin** content can fluctuate from 3 to 20% depending on cultivars and location (rhizome, lateral roots) and its composition varies with the chemotype.

KAVA, Pharmacological research

- The pyrones
 - induce sleep in rodents (per os) and are sedatives in rodents, cats, and rabbits.
 - cause muscle relaxation and several are anticonvulsant (strychnine, electroshock).
- The **kavapyrones** (DHK, DHM) are analgesics and weak local anesthetics.
- The aqueous extract and the lipid-soluble fraction decrease spontaneous movement, but
- the (mild) sedation induced by the aqueous extract is not accompanied by a loss of muscular tone;
- the resin induces sleep, but the aqueous extract does not (mouse, IP).
- Kava and kawain induce sleep by acting on the limbic system (EEG in cats)
- Klinical trials indicated that a kava extract is more efficacious than a placebo in patients who suffer from non-psychotic anxiety.

KAVA, Uses

- In Germany, pharmaceuticals based on standardized extracts (i.e., 35-120 mg kavapyrones)
 were promoted as sleep disorder and anxiety medicines.
- Nowadays their application is contraindicated due to the
- hepatotoxic adverse reactions reported in association with the use of all types of kava products in the South Pacific Islands, Australia, Europe, and the US.
- It appears that **poor quality of the kava material** was responsible for the **liver toxicity**.
- Therefore, a sophisticated approach to establish kava quality standardizations is needed for safe human use of kava as
 - relaxing traditional beverages,
 - the anxiolytic drugs, and
 - recreational dietary supplements.